

# Porphyrin Cosensitization for a Photovoltaic Efficiency of 11.5%: A Record for Non-Ruthenium Solar Cells Based on Iodine Electrolyte

Yongshu Xie,\* Yunyu Tang, Wenjun Wu, Yueqiang Wang, Jingchuan Liu, Xin Li, He Tian, and Wei-Hong Zhu\*

Key Laboratory for Advanced Materials and Institute of Fine Chemicals, East China University of Science and Technology, Meilong Road 130, Shanghai 200237, People's Republic of China

**Supporting Information** 

ABSTRACT: Dye-sensitized solar cells (DSSCs) are promising for utilizing solar energy. To achieve high efficiencies, it is vital to synergistically improve the photocurrent  $(I_{sc})$  and the photovoltage  $(V_{oc})$ . In this respect, conjugation framework extension and cosensitization are effective for improving the absorption and the  $J_{sc}$ which, however, is usually accompanied by undesirably decreased V<sub>oc</sub>. Herein, based on a rationally optimized porphyrin dye, we develop a targeted coadsorption/ cosensitization approach for systematically improving the  $V_{\rm oc}$  from 645 to 727, 746, and 760 mV, with synergistical  $J_{sc}$  enhancement from 18.83 to 20.33 mA cm<sup>-2</sup>. Thus, the efficiency has been dramatically enhanced to 11.5%, which keeps the record for nonruthenium DSSCs using the  $I_2/I_3^$ electrolyte. These results compose an alternative approach for developing highly efficient DSSCs with relatively high  $V_{\rm oc}$  using traditional iodine electrolyte.

I ncreasing energy demand and environmental pollution problems have prompted research on clean and inexhaustive energy sources.<sup>1</sup> In this respect, utilization of solar energy has been demonstrated to be highly promising.<sup>2</sup> On the basis of traditional silicon-based solar cells, dye-sensitized solar cells (DSSCs) present a promising potential because of their low production cost, easy fabrication, and relatively high solar energy conversion efficiency.<sup>3</sup> Since the first report in 1991, increasing interests have been focused on improving the efficiencies of DSSCs.<sup>4</sup> To this end, it is vital to simultaneously improve the photocurrent ( $J_{sc}$ ) and the photovoltage ( $V_{oc}$ ).

Porphyrin dyes have been widely exploited in DSSCs due to their easily modulated structures and strong absorption bands in a wide wavelength range, except the NIR region and around 500 nm.<sup>5,6</sup> To further improve sunlight harvesting in these two regions, a combined approach of extending the porphyrin conjugation frameworks and application of a cosensitizer is effective for achieving a panchromatic absorption and a high  $J_{\rm sc}$ .<sup>7,8</sup> However, extended conjugation structures may undesirably result in decreased  $V_{\rm oc}$  because of aggravated dye aggregation,<sup>9,10</sup> and cosensitization also usually results in undesirably decreased or nearly unaffected  $V_{\rm oc}$  with only a few exceptions.<sup>11</sup> Herein, we focus on a systematic cosensitization approach for synergistically improving the  $V_{\rm oc}$  and  $J_{\rm sc}$  thus pursuing high efficiencies for DSSCs based on novel extended porphyrin dyes. In virtue of the delicately designed coadsorption/cosensitization, the  $V_{\rm oc}$  for DSSCs based on XW11 (Figure 1) was stepwise improved from 645 to 727, 746, and 760 mV, with a simultaneous  $J_{sc}$ 



Figure 1. Molecular structures of the porphyrin dyes (XW9, XW10, and XW11), the coadsorbent (CDCA), and the cosensitizers (C1 and WS-5).

enhancement from 18.83 to 20.33 mA cm<sup>-2</sup>. As a consequence, the targeted cosensitization can even achieve a photovoltaic efficiency of 11.5%, to the best of our knowledge, a record for nonruthenium DSSCs based on iodine electrolyte.

Sensitizers with strong absorption are highly desirable for improving light harvesting efficiency and  $J_{sc}$ . In this regard, porphyrins show a strong Soret band (400–450 nm) and moderate Q-bands (550–600 nm), covering the visible to the near-IR region, with remarkably high molecular absorption coefficients in the visible region (up to  $10^5 \text{ M}^{-1} \text{ cm}^{-1}$ ).<sup>5,6</sup> The incorporation of conjugated groups can further extend the light

Received: September 21, 2015 Published: October 22, 2015

# Journal of the American Chemical Society

response range. However, the conjugation extension may induce severe dye aggregation, resulting in a decreased  $V_{oc}$ . Thus, all three dyes were specifically wrapped with long alkoxy chains.<sup>12</sup> Furthermore, a phenothiazine moiety was introduced as the electron donor, given its strong electron-donating character and a nonplanar butterfly conformation favorable for further suppressing the aggregation.<sup>13</sup>

As expected, all the dyes feature typical porphyrin absorption spectra with an intense Soret band within 400–480 nm and less intense Q bands within 550–730 nm (Figure 2, Table S1).



Figure 2. Absorption spectra of dyes XW9, XW10, XW11, C1, and WS-5 in THF.

Compared with XW9, the additional ethynylene bridge and the auxiliary benzothiadiazole group in XW10 and XW11 red-shifted the absorption onset wavelengths from 650 to 700 and 730 nm, respectively, which is favorable for harvesting the sunlight.<sup>14</sup> Furthermore, broadened absorption bands were observed on  $TiO_2$  films (Figure S1) with respect to the corresponding solution spectra, which is also favorable for sunlight harvesting.

The trend of successively red-shifted absorption for XW9, XW10, and XW11 is well consistent with that obtained from the density functional theory (DFT) calculations using the Gaussian 09 program package (Figure S2, Table S2),<sup>15</sup> which also revealed that electron densities of the HOMO orbitals are mainly delocalized over the donor and the porphyrin framework, while the LUMO orbitals are predominantly delocalized over the anchoring group and the porphyrin framework. Thus, the electron transfer from the HOMO to the LUMO can be easily accessible in the electron redistribution from the donor to the anchoring moiety, enabling electron injection from the LUMO to the conduction band of TiO2. Moreover, suitable HOMO and LUMO energy levels are vital for the electron injection and dye regeneration processes. The HOMOs were obtained to be 0.83, 0.72, and 0.73 V for XW9, XW10, and XW11, respectively, which are more positive than the  $I_2/I_3^-$  redox potential (0.4 V vs NHE) (Figures S3 and S4, Table S3), and the corresponding LUMO levels were obtained to be -1.12, -1.13, and -1.05 V, respectively, which are more negative than the conduction band edge (CB) of  $TiO_2$  (-0.5 V vs NHE). These results are indicative that both the dye regeneration and electron injection processes are thermodynamically feasible.<sup>16</sup>

The porphyrin dyes were evaluated as the sensitizers of DSSCs based on an  $I_2/I_3^-$  electrolyte. Figure 3 shows the photocurrent–voltage (J-V) curves of the DSSCs measured under simulated AM 1.5G irradiation (100 mW cm<sup>-2</sup>), and the corresponding data are listed in Table 1, with more details listed in Tables S4 and S5. The solar-to-electric power conversion efficiencies (PCE) lie within 7.8–8.6%. As mentioned above, XW9, XW10, and XW11 have successively extended conjugation frameworks



**Figure 3.** Photocurrent–voltage (J-V) characteristics (a,c,e) and IPCE action spectra (b,d,f) for the DSSCs based on **XW9**, **XW10**, and **XW11**, and cosensitized with **C1** or **WS-5**.

Table 1. Optimized Photovoltaic Data for DSSCs based on XW9, XW10, and XW11 under AM 1.5G Illumination (100 mW  $cm^{-2}$ )

dyes	$J_{\rm sc}~({\rm mA~cm^{-2}})$	$V_{\rm oc}~({\rm mV})$	FF	PCE (%)
XW9	$16.27\pm0.11$	$728 \pm 4$	$0.698 \pm 0.004$	$8.2 \pm 0.1$
XW10	$17.90\pm0.04$	$711 \pm 5$	$0.684 \pm 0.005$	$8.6\pm0.1$
XW11	$18.83 \pm 0.28$	$645 \pm 3$	$0.642 \pm 0.003$	$7.8\pm0.1$
XW9 + CDCA	$16.17\pm0.13$	$740 \pm 3$	$0.689 \pm 0.005$	$8.2 \pm 0.1$
XW10 + CDCA	$17.51 \pm 0.17$	739 ± 4	$0.680 \pm 0.005$	$8.8 \pm 0.1$
XW11 + CDCA	$18.26 \pm 0.27$	727 ± 2	$0.701 \pm 0.004$	$9.3 \pm 0.1$
XW9 + C1	$17.01\pm0.12$	$764 \pm 2$	$0.718 \pm 0.001$	$9.3 \pm 0.1$
XW10 + C1	$18.24 \pm 0.08$	$753 \pm 4$	$0.742 \pm 0.003$	$10.1\pm0.2$
XW11 + C1	$19.52\pm0.05$	746 ± 5	$0.740\pm0.002$	$10.6 \pm 0.1$
XW9 + WS-5	$17.70 \pm 0.16$	$770\pm2$	$0.741 \pm 0.003$	$10.1 \pm 0.1$
XW10 + WS-5	$19.01\pm0.11$	$765 \pm 3$	$0.764 \pm 0.004$	$11.0\pm0.2$
XW11 + WS-5	$20.33 \pm 0.21$	$760 \pm 4$	$0.744 \pm 0.004$	$11.5\pm0.2$

and absorption ranges. As a result, they show increasing  $J_{sc}$  of 16.27, 17.90, and 18.83 mA cm<sup>-2</sup> (Figure 3a), respectively. Observation of the highest  $J_{sc}$  for XW11 is consistent with its broadest absorption. On the other hand, the extended conjugation frameworks brought forth a decreasing  $V_{oc}$  of 728, 711, and 645 mV for XW9, XW10, and XW11, respectively, which might be ascribed to the detrimental dye aggregation. As a result of the contradictory trends of  $J_{sc}$  and  $V_{oc}$ , XW9, XW10, and XW11 demonstrate DSSC efficiencies of 8.2, 8.6, and 7.8%, respectively, with the highest photovoltaic efficiency achieved by XW10. It is also noteworthy that the efficiencies for XW9 and XW10 are higher than those of the corresponding carbazole-based dyes,<sup>17</sup> indicating that the electron-donating effect of phenothiazine is superior to carbazole.

Based on the photovoltaic data, we checked the incidentphoton-to-current conversion efficiency (IPCE) action spectra to understand the contribution of absorption at different wavelengths to the  $J_{sc}$  (Figure 3). All the three dyes show a broad IPCE plateau in the visible range. Compared with **XW9**, the insertion of an ethynylene bridge in **XW10** red-shifted the onset wavelength of photocurrent response from 730 to 775 nm, resulting in an increase of  $J_{sc}$  from 16.27 to 17.90 mA cm<sup>-2</sup>. Further introduction of the benzothiadiazole group in **XW11** induced an impressive broad IPCE action spectrum with the onset wavelength red-shifted to 830 nm, resulting in a highest  $J_{sc}$  of 18.83 mA cm<sup>-2</sup>.

Generally, the utilization of a coadsorbent is effective for suppressing dye aggregation, thus enhancing the  $V_{\rm oc}$ . In this respect, chenodeoxycholic acid (CDCA, Figure 1) is the most commonly used due to its nonplanar and bulky configuration.<sup>18</sup> Given the undesirably low  $V_{oc}$  values of **XW9–XW11**, it may be feasible to improve the efficiencies by enhancing the  $V_{oc}$  with a coadsorbent. Thus, DSSCs were fabricated by dipping the  $TiO_2$ films into the dye solutions containing CDCA. Upon using 5 mM CDCA, the  $V_{\rm oc}$  for XW9, XW10, and XW11 were improved from 728, 711, and 645 mV to 753, 750, and 738 mV, respectively (Tables S6 and S7, Figure S6). Provided that  $\Delta V_{oc}$  lies in the sequence of XW9 < XW10 < XW11, the suppression of dye aggregation is the most pronounced for XW11 arising from the most serious aggregation due to its largest conjugation framework. Accompanied with the increased  $V_{\rm oct}$  the corresponding  $J_{sc}$  values were undesirably decreased to 15.16, 16.47, and 17.17 mA cm<sup>-2</sup>, respectively, which is related to the decreased coverage of porphyrin dyes on the TiO<sub>2</sub> films upon coadsorption of CDCA, thus leading to a loss in light harvesting and a decrease in  $J_{sc}$ . As a result, the enhancement in photovoltaic efficiency is neglectable for XW9 and XW10. Only XW11 shows a moderate efficiency enhancement from 7.8% to 9.3%, which mainly results from the dramatic enhancement in the  $V_{\rm oc}$ .

Although CDCA can be used for improving the  $V_{oc}$  by suppressing the dye aggregation, it cannot absorb the sunlight, and thus it does not have photovoltaic effect, which causes decreased  $J_{sc}$  and thus is unfavorable for dramatic improving the photovoltaic efficiency. Hence, we continued to delicately design the cosensitization approach for pursuing the synergistic improvement of  $V_{oc}$  and  $J_{sc}$ . Considering the fact that C1 (Figure  $(1)^{17}$  demonstrates a broad absorption peak around 500 nm (Figure 2), which well compensates the absorption valley of the porphyrin dyes in this region, we initially used C1 for cosensitization. As expected, the IPCE valleys of the porphyrin dyes around 500 nm were indeed filled up, with the values remaining higher than 75% in this region (Figure 3d,f). As a result, the cosensitized cells show the enhanced  $J_{sc}$  values  $(17.01-19.52 \text{ mA cm}^{-2})$  with respect to corresponding individual porphyrin-sensitized cells (Figure 3c, Table 1). More strikingly, the  $V_{oc}$  values for XW9, XW10, and XW11 upon cosensitization with C1 were simultaneously improved from 728, 711, and 645 mV to 764, 753, and 746 mV (Table 1), respectively. As we know, a typical cosensitized solar cell usually affords a  $V_{\rm oc}$  value between those obtained for the individual dyes.<sup>17</sup> Hence, the cosensitized cells show  $V_{\rm oc}$  values between those for individual porphyrin dyes (728-645 mV) and C1 (780 mV). Through successful synergistic enhancement of  $J_{sc}$  and  $V_{oc}$ , the high photovoltaic efficiencies of 9.3, 10.1, and 10.6% were achieved by cosensitization of C1 with XW9, XW10, and XW11, respectively (Tables 1, S8, and S9).

Based on the successful application of cosensitizer C1, it is anticipated that the efficiencies may be further improved if another matchable cosensitizer with a higher  $V_{oc}$  and a higher efficiency is utilized. Thus, we continued to utilize WS-5 as the alternative cosensitizer, which individually demonstrated a high  $V_{\rm oc}$  of 791 mV and a high efficiency of 8.38%.<sup>19</sup> As expected, the  $V_{\rm oc}$  values of XW9–XW11 were further improved to 770–760 mV upon cosensitization with WS-5 (Tables 1, S10, and S11). Meanwhile, the IPCE curves showed a high plateau with the maximum reaching as high as 90%, thus achieving high  $J_{sc}$  of 17.70–20.33 mA  $cm^{-2}$  (Figure 3e,f). The 20.33 mA  $cm^{-2}$ achieved for cosensitization of XW11 with WS-5 (XW11 + WS-5) is slightly higher than the value integrated from the IPCE spectrum (19.80 mA cm<sup>-2</sup>, Figure S7). Similar observations have also been reported for a number of porphyrin dyes,<sup>20</sup> which may be caused by the more efficient charge transport and collection<sup>2</sup> as well as stronger thermal effect associated with the full sunlight irradiation.<sup>22</sup> As a result, the high photovoltaic efficiencies of 10.1-11.5% were achieved. The excellent photovoltaic behavior of DSSCs based on cosensitization of XW11 with WS-5 was further tested by fabricating a total of 20 cells, which demonstrated good reproducibility (Table S11). Furthermore, one of the cells was also sent to the National Institute of Metrology (NIM), China for certification, and a certified efficiency of 10.9% was obtained (Figure S8). In addition to the high efficiency, the DSSCs also showed satisfactory photostability, with the efficiency remaining at 86% of the initial value after 1000 h of visible-light soaking (Figure S9).

As described above, the  $V_{\rm oc}$  values of **XW9–XW11** can be systematically improved through cosensitization. Generally, the alteration of the photovoltage originates from a shift of the TiO<sub>2</sub> electron quasi-Fermi-level, which may be ascribed to two major reasons: (i) a shift in the TiO<sub>2</sub> conduction band edge ( $E_{\rm CB}$ ), which can be inferred from the chemical capacitance ( $C_{\mu}$ ), and (ii) a fluctuation of electron density, which is related to the electron lifetime ( $\tau$ ) determined by the charge recombination rate.<sup>Sa,19</sup> Thus, the electrochemical impedance spectroscopy was checked to obtain the corresponding  $C_{\mu}$  and  $\tau$ .

For XW9, the individual porphyrin-based solar cells and the cosensitized ones showed similar  $C_{\mu}$  values (Figure S10a). Whereas, the corresponding electron lifetimes lie in the order of XW9 < XW9 + C1 < XW9 + WS-5 (Figure S10b), which is indicative of the same order of decreasing charge recombination rates and consistent with the sequence of increasing  $V_{oc}$ . These results indicate that the  $V_{oc}$  of XW9-based DSSCs are governed by the charge recombination rates. For XW10, the trends for both  $C_{\mu}$  and  $\tau$  agree well with that of the  $V_{oc}$  values (Figure S10c,d), suggesting that the  $V_{oc}$  of XW10-based DSSC may be related to both the charge recombination rates and the positions of the conduction band. Similar to XW9, the  $V_{oc}$  values of XW11-based DSSCs are governed by the charge recombination rates (Figure S10e,f).<sup>23</sup>

In summary, using a phenothiazine-based electron donor, three porphyrin dyes (**XW9–XW11**) were rationally designed. Ethynylene and benzothiadiazole (BTD) units were successively introduced to tailor the absorption to longer wavelengths, with the onset wavelengths of photocurrent response red-shifted from 730 nm (**XW9**) to 830 nm (**XW11**), which is quite exceptional for organic sensitizers, and thus the corresponding  $J_{\rm sc}$  was improved from 16.27 to 18.83 mA cm<sup>-2</sup>. However, the  $V_{\rm oc}$  value was undesirably decreased from 728 to 645 mV due to the more severe dye aggregation caused by the larger conjugation framework of **XW11**. To improve the  $V_{\rm oc}$  and the photovoltaic efficiency, a delicately designed coadsorption/cosensitization approach was utilized to systematically improve the  $V_{\rm oc}$  from 645 to 727, 746, and 760 mV, which was accompanied by synergistically enhanced  $J_{\rm sc}$  to 20.33 mA cm<sup>-2</sup>. Finally, the

efficiencies for **XW11** were successfully improved from 7.8% to 11.5%, which keeps the record for nonruthenium DSSCs using the  $I_2/I_3^-$  electrolyte. These results provide further insight into developing efficient DSSCs with synergistically enhanced photovoltage and photocurrent.

# ASSOCIATED CONTENT

### **Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b09665.

Synthetic procedures and characterization data of all new compounds; details for all physical characterizations (PDF)

## AUTHOR INFORMATION

# **Corresponding Authors**

\*yshxie@ecust.edu.cn

\*whzhu@ecust.edu.cn

#### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

This work was supported by the Science Fund for Creative Research Groups (21421004), Distinguished Young Scholars (21325625), NSFC/China (21472047, 91227201), and the Oriental Scholarship

# REFERENCES

(1) (a) O'Regan, B.; Grätzel, M. *Nature* **1991**, *353*, 737. (b) Higashino, T.; Imahori, H. *Dalton Trans.* **2015**, *44*, 448. (c) Imahori, H.; Umeyama, T.; Ito, S. *Acc. Chem. Res.* **2009**, *42*, 1809.

(2) (a) Hagfeldt, A.; Boschloo, G.; Sun, L.; Kloo, L.; Pettersson, H. Chem. Rev. 2010, 110, 6595. (b) Qu, S. Y.; Hua, J. L.; T ian, H. Sci. China: Chem. 2012, 55, 677. (c) Mishra, A.; Fischer, M. K. R.; Bäuerle, P. Angew. Chem., Int. Ed. 2009, 48, 2474. (d) Wu, Y. Z.; Zhu, W. H.; Zakeeruddin, S. M.; Grätzel, M. ACS Appl. Mater. Interfaces 2015, 7, 9307.

(3) (a) Zhang, S.; Yang, X.; Numata, Y.; Han, L. Y. Energy Environ. Sci. 2013, 6, 1443. (b) Chou, C. C.; Hu, F. C.; Yeh, H. H.; Wu, H. P.; Chi, Y.; Clifford, J. N.; Palomares, E.; Liu, S. H.; Chou, P. T.; Lee, G. H. Angew. Chem., Int. Ed. 2014, 53, 178. (c) Ishida, M.; Hwang, D.; Zhang, Z.; Choi, Y. J.; Oh, J.; Lynch, V. M.; Kim, D. Y.; Sessler, J. L.; Kim, D. ChemSusChem 2015, 8, 2967.

(4) (a) Wu, Y. Z.; Zhu, W. H. Chem. Soc. Rev. **2013**, *42*, 2039. (b) Yang, J. B.; Ganesan, P.; Teuscher, J.; Moehl, T.; Kim, Y. J.; Yi, C.; Comte, P.; Pei, K.; Holcombe, T. W.; Nazeeruddin, M. K.; Hua, J. L.; Zakeeruddin, S. M.; Tian, H.; Grätzel, M. J. Am. Chem. Soc. **2014**, *136*, 5722. (c) Yao, Z. Y.; Zhang, M.; Wu, H.; Yang, L.; Li, R. Z.; Wang, P. J. Am. Chem. Soc. **2015**, *137*, 3799. (d) Yao, Z. Y.; Zhang, M.; Li, R. Z.; Yang, L.; Qiao, Y. N.; Wang, P. Angew. Chem., Int. Ed. **2015**, *54*, 5994.

(5) (a) Li, L. L.; Diau, E. W. G. Chem. Soc. Rev. 2013, 42, 291.
(b) Mathew, S.; Yella, A.; Gao, P.; Baker, R. H.; Curchod, B. F. E.; Astani, N. A.; Tavernelli, I.; Rothlisberger, U.; Nazeeruddin, M. K.; Grätzel, M. Nat. Chem. 2014, 6, 242. (c) Higashino, T.; Fujimori, Y.; Sugiura, K.; Tsuji, Y.; Ito, S.; Imahori, H. Angew. Chem., Int. Ed. 2015, 54, 9052.
(d) Mai, C. L.; Moehl, T.; Hsieh, C. H.; Décoppet, J. D.; Zakeeruddin, S. M.; Grätzel, M.; Yeh, C. Y. ACS Appl. Mater. Interfaces 2015, 7, 14975.

(6) Pelleja, L.; Kumar, C. V.; Clifford, J. N.; Palomares, E. J. Phys. Chem. C **2014**, *118*, 16504.

(7) (a) Wang, C. L.; Shiu, J. W.; Hsiao, Y. N.; Chao, P. S.; Diau, E. W.
G.; Lin, C. Y. J. Phys. Chem. C 2014, 118, 27801. (b) Yao, Z. Y.; Wu, H.;
Ren, Y. M.; Guo, Y. C.; Wang, P. Energy Environ. Sci. 2015, 8, 1438.
(8) Kang, S. H.; Choi, I. T.; Kang, M. S.; Eom, Y. K.; Ju, M. J.; Hong, J.

(6) Kang, 9. 11., Choi, 1. 1., Kang, M. S., Eoni, T. K., Ju, M. J., Hong, J. Y.; Kang, H. S.; Kim, H. Y. J. Mater. Chem. A **2013**, *1*, 3977.

(9) (a) Joly, D.; Pellejà, L.; Narbey, S.; Oswald, F.; Meyer, T.; Kervella, Y.; Maldivi, P.; Clifford, J. N.; Palomares, E.; Demadrille, R. *Energy*  *Environ. Sci.* **2015**, *8*, 2010. (b) Yan, C. C.; Ma, W. T.; Ren, Y. M.; Zhang, M.; Wang, P. ACS Appl. Mater. Interfaces **2015**, *7*, 801.

(10) Cabau, L.; Kumar, C. V.; Moncho, A.; Clifford, J. N.; López, N.; Palomares, E. *Energy Environ. Sci.* **2015**, *8*, 1368.

(11) Lan, C. M.; Wu, H. P.; Pan, T. Y.; Chang, C. W.; Chen, C. T.;
Wang, C. L.; Lin, C. Y.; Diau, E. W. G. *Energy Environ. Sci.* 2012, *5*, 6460.
(12) (a) Sanchis, T. R.; Guo, B. C.; Wu, H. P.; Pan, T. Y.; Lee, H. W.;
Raga, S. R.; Santiago, F. F.; Bisquert, J.; Yeh, C. Y.; Diau, E. W. G. *Chem. Commun.* 2012, *48*, 4368. (b) Wang, C. L.; Lan, C. M.; Hong, S. H.;
Wang, Y. F.; Pan, T. Y.; Chang, C. W.; Kuo, H. H.; Kuo, M. Y.; Diau, E.
W. G.; Lin, C. Y. *Energy Environ. Sci.* 2012, *5*, 6933.

(13) Hua, Y.; Chang, S.; Huang, D. D.; Zhou, X.; Zhu, X. J.; Zhao, J. Z.; Chen, T.; Wong, W. Y.; Wong, W. K. *Chem. Mater.* **2013**, *25*, 2146.

(14) Zhu, W. H.; Wu, Y. Z.; Wang, S. T.; Li, W. Q.; Li, X.; Chen, J.; Wang, Z. S.; Tian, H. Adv. Funct. Mater. 2011, 21, 756.

(15) (a) Becke, A. D. J. Chem. Phys. 1993, 98, 5648. (b) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, N. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, Revision A.2, Gaussian, Inc.: Wallingford, CT, 2009.

(16) Hagberg, D. P.; Yum, J. H.; Lee, H.; Angelis, F. D.; Marinado, T.; Karlsson, K. M.; Baker, R. H.; Sun, L. C.; Hagfeldt, A.; Grätzel, M.; Nazeeruddin, M. K. *J. Am. Chem. Soc.* **2008**, *130*, 6259.

(17) (a) Wang, Y. Q.; Chen, B.; Wu, W. J.; Li, X.; Zhu, W. H.; Tian, H.; Xie, Y. S. Angew. Chem., Int. Ed. **2014**, 53, 10779. (b) Li, Z.; Li, Q. Q. Sci. China: Chem. **2014**, 57, 1491. (c) Wei, T. T.; Sun, X.; Li, X.; Ågren, H.; Xie, Y. S. ACS Appl. Mater. Interfaces **2015**, 7, 21956.

(18) (a) Mikroyannidis, J. A.; Suresh, P.; Roy, M. S.; Sharma, G. D. *Electrochim. Acta* **2011**, *56*, 5616. (b) Lim, J.; Kwon, Y. S.; Park, T. *Chem. Commun.* **2011**, *47*, 4147.

(19) Li, W. Q.; Wu, Y. Z.; Zhang, Q.; Tian, H.; Zhu, W. H. ACS Appl. Mater. Interfaces **2012**, *4*, 1822.

(20) (a) Wu, H. P.; Qu, Z. W.; Pan, T. Y.; Lan, C. M.; Huang, W. K.; Lee, H. W.; Reddy, N. M.; Chen, C. T.; Chao, W. S.; Yeh, C. Y.; Diau, E. W. G. *Energy Environ. Sci.* **2012**, *5*, 9843. (b) Chang, S.; Wang, H. D.; Hua, Y.; Li, Q.; Xiao, X. D.; Wong, W. K.; Wong, W. Y.; Zhu, X. J.; Chen, T. J. Mater. Chem. A **2013**, *1*, 11553.

(21) Frank, A. L.; Kopidakis, N.; Lagemaat, J. Coord. Chem. Rev. 2004, 248, 1165.

(22) Cao, Y. M.; Bai, Y.; Yu, Q. J.; Cheng, Y. M.; Liu, S.; Shi, D.; Gao, F. F.; Wang, P. J. Phys. Chem. C 2009, 113, 6290.

(23) (a) Yella, A.; Lee, H. W.; Tsao, H. N.; Yi, C. Y.; Chandiran, A. K.; Nazeeruddin, M. K.; Diau, E. W. G.; Yeh, C. Y.; Zakeeruddin, S. M.; Grätzel, M. Science **2011**, 334, 629. (b) Hardin, B. E.; Snaith, H. J.; McGehee, M. D. Nat. Photonics **2012**, 6, 162.